Structure and Functionality of Supramolecules **Constructed by Amphiphiles**

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The surface pressure-area $(\pi$ -A) isotherms of dioctadecyldimethylammonium chloride $(2C_{18}DAC)$ and dioctadecyldimethylammonium cinnamate $(2C_{18}DA \cdot Cin)$ were measured, and their time dependence was examined. The surfaces of Langmuir-Blodgett films of 2C₁₈DAC and 2C₁₈DA · Cin were observed by atomic force microscopy (AFM). Then the location and effect of cinnamate anions for molecular organization on Langmuir monolayer of 2C₁₈DA were discussed. Just after the preparation, cinnamate ions bind on hydrophilic surface of $2C_{18}DA$ monolayer, by forming ion pairs with $2C_{18}DA$ cations and by directing aromatic groups toward the interior of water subphase. When Langmuir film is maintained during long time at a constant surface pressure, cinnamate ions are intercalated into monolayer, since aromatic groups penetrate into hydrophobic interior of monolayer.